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Hydrothermally prepared mansfieldite, AlAsO₄·2H₂O (aluminum arsenate dihydrate), contains a vertex-sharing three-dimensional network of *cis*-AlO₄(H₂O)₂ octahedra and AsO₄ tetrahedra [$d_{\text{av}}(\text{Al}-\text{O}) = 1.907(2)$ Å, $d_{\text{av}}(\text{As}-\text{O}) = 1.685(2)$ Å and $\theta_{\text{av}}(\text{Al}-\text{O}-\text{As}) = 134.5(1)^\circ$].

Comment

On the basis of cell parameters, symmetry and composition, the present phase is a synthetic analogue of mansfieldite, AlAsO₄·2H₂O (Allen *et al.*, 1948; Ronis & D'Yvoire, 1969). Mansfieldite is a member of the $M^{\text{III}}\text{XO}_4\cdot 2\text{H}_2\text{O}$ ($M = \text{Al, Fe, Ga, Cr; X = P, As}$) family of phases exemplified by variscite, AlPO₄·2H₂O (Kniep *et al.*, 1977).

The *cis*-AlO₄(H₂O)₂ octahedron [$d_{\text{av}}(\text{Al}-\text{O}) = 1.907(2)$ Å] and AsO₄ tetrahedron [$d_{\text{av}}(\text{As}-\text{O}) = 1.685(2)$ Å] share vertices, as Al—O—As bonds [atoms O1—O4 with $\theta_{\text{av}} = 134.5(1)^\circ$], resulting in a three-dimensional network of polyhedra. Three of the four water H atoms participate in O—H···O hydrogen bonds in a similar manner to that seen in variscite (Kniep *et al.*, 1977).

Experimental

A starting mixture of KOH (1 M, 4 ml), Al(NO₃)₃ (0.5 M, 8 ml) and H₃AsO₄ (1 M, 4 ml) was heated to 443 K in a 23-ml-capacity teflon-lined hydrothermal bomb for 4 d. Upon cooling the bomb to ambient temperature over a period of 2–3 h, a crop of small (to 0.1 mm) blocks and octahedra of the title compound was recovered by vacuum filtration and washing with water and acetone.

Crystal data

AlAsO₄·2H₂O
 $M_r = 201.93$
Orthorhombic, *Pbca*
 $a = 8.8218(5)$ Å
 $b = 9.8252(6)$ Å
 $c = 10.1163(6)$ Å
 $V = 876.8(2)$ Å³
 $Z = 8$
 $D_x = 3.06$ Mg m⁻³

Mo $K\alpha$ radiation
Cell parameters from 2390
reflections
 $\theta = 3.70\text{--}30.66^\circ$
 $\mu = 7.89$ mm⁻¹
 $T = 298$ K
Faceted block, colourless
0.08 × 0.08 × 0.08 mm

Data collection

Bruker SMART1000 CCD area-detector diffractometer
 ω scans
Absorption correction: multi-scan (SADABS; Bruker, 1999)
 $T_{\min} = 0.470$, $T_{\max} = 0.532$
8491 measured reflections

1587 independent reflections
1059 reflections with $I > \sigma(I)$
 $R_{\text{int}} = 0.039$
 $\theta_{\text{max}} = 32.5^\circ$
 $h = -12 \rightarrow 9$
 $k = -14 \rightarrow 11$
 $l = -14 \rightarrow 11$

Refinement

Refinement on F
 $R = 0.025$
 $wR = 0.026$
 $S = 1.047$
1059 reflections
90 parameters
All H-atom parameters refined

Weighting: Chebychev polynomial with 3 parameters (Carruthers & Watkin, 1979): 0.139, 0.174, 0.091
 $(\Delta/\sigma)_{\text{max}} = 0.000112$
 $\Delta\rho_{\text{max}} = 0.73$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.68$ e Å⁻³

Table 1
Selected geometric parameters (Å, °).

Al1—O1	1.903 (2)	Al1—O6	1.977 (3)
Al1—O2 ⁱ	1.883 (3)	As1—O1	1.684 (2)
Al1—O3 ⁱⁱ	1.894 (2)	As1—O2	1.679 (2)
Al1—O4 ⁱⁱⁱ	1.866 (3)	As1—O3	1.685 (2)
Al1—O5	1.920 (3)	As1—O4	1.692 (2)
Al1—O1—As1	133.20 (13)	Al1 ^v —O3—As1	135.42 (14)
Al1 ^{iv} —O2—As1	135.50 (15)	Al1 ⁱⁱⁱ —O4—As1	133.99 (14)

Symmetry codes: (i) $-x, \frac{1}{2} + y, \frac{1}{2} - z$; (ii) $\frac{1}{2} - x, \frac{1}{2} + y, z$; (iii) $-x, -y, -z$; (iv) $-x, y - \frac{1}{2}, \frac{1}{2} - z$; (v) $\frac{1}{2} - x, y - \frac{1}{2}, z$.

Table 2
Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
O5—H51···O4 ⁱ	0.82 (8)	1.78 (8)	2.592 (3)	169 (8)
O5—H52···O1 ⁱⁱ	0.72 (6)	2.04 (6)	2.729 (3)	160 (6)
O6—H62···O3	0.94 (7)	1.70 (7)	2.578 (4)	155 (7)

Symmetry codes: (i) $-x, \frac{1}{2} + y, \frac{1}{2} - z$; (ii) $\frac{1}{2} + x, y, \frac{1}{2} - z$.

Data collection: SMART (Bruker, 1999); cell refinement: SMART; data reduction: SMART; program(s) used to refine structure: CRYSTALS (Watkin *et al.*, 1997); software used to prepare material for publication: CRYSTALS.

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